## IN THE CLAIMS

- 1. (Currently Amended) A process for the reaction of a ketone or an aldehyde with a phenol to form a bisphenol comprising reacting a feed comprising a phenol, water and a ketone or aldehyde and water in the presence of an ion exchange resin catalyst to produce an effluent; determining the para-para bisphenol selectivity of the reaction; and adjusting the concentration of the water in the feed based upon the para-para bisphenol selectivity
- 2. (Original) The process of Claim 1, wherein said ketone is acetone and said phenol is a hydroxy aromatic compound with at least one unsubstituted position.
- 3. (Original) The process of Claim 2 wherein the unsubstituted position is para to the hydroxy position.
- 4. (Original) The process of Claim 2 wherein the phenol is substituted with at least one inert substituent.
- 5. (Original) The process of Claim 1, wherein the phenol to ketone mole ratio is about 4 to about 65.
- 6. (Currently amended) The process of Claim 1, wherein said phenol emprises is selected from the group consisting of phenol, 2-cresol, 3-cresol, 2,6-dimethylphenol, resorcinol, napthol or and mixtures of two or more of the foregoing phenols.
- 7. (Currently amended) The process of Claim 1, wherein said ketone emprises is selected from the group consisting of 9-fluorenone, benzophenone, acetone, acetophenone, cyclohexanone, 3,3,5-trimethylcyclohexanone, 4-hydroxyacetophenone, 4,4'-dihydroxybenzophenone er aand mixtures of two or more of the foregoing ketones.
- 8. (Currently amended) The process of Claim 1, wherein said ketone-aldehyde eemprises is selected from the group consisting of formaldehyde, acetaldehyde, propionaldehyde, butyraldehyde er aand mixtures of two or more of the foregoing aldehydes.

- 9. (Original) The process of Claim 1, wherein the reaction has a weight hour space velocity of about 0.1 to about 10.
- 10. (Original) The process of Claim 1, wherein the feed has a temperature of about 40 to about 90 °C.
- 11. (Original) The process of Claim 1, wherein the para/para selectivity of the reaction is greater than about 94%.
- 12. (Original) The process of Claim 1, wherein the catalyst has a degree of crosslinking of about 1.5% to about 6.0%.
- 13. (Original) The process of Claim 12, wherein catalyst is a mixture of resins of different degrees of crosslinking wherein the degree of crosslinking of each resin comprising the mixture is about 1.5 to about 6%.
- 14. (Original) The process of Claim 1 wherein the catalyst has a degree of neutralization of about 35 to about 60 mole %.
- 15. (Original) The process of Claim 1, wherein the process has a conversion level greater than or equal to about 70%.
- 16. (Original) The process of Claim 1, wherein the amount of water in the feed is less than or equal to about 5 weight percent, based on the total weight of the feed.
- 17. (Original) The process of Claim 1, wherein the catalyst further comprises an attached promoter.
- 18. (Original) The process of Claim 17, wherein the attached promoter is cysteamine, 4-pyridylethylmercaptan or a combination of the foregoing.
- 19. (Currently amended) The process of Claim 1, wherein the ketone or aldehyde is added in a single portion.

- 20. (Currently amended) The process of Claim 1, wherein the ketone or aldehyde is added in multiple portions.
- 21. (Original) The process of Claim 1, wherein the feed comprises recycled compounds.
- 22. (Original) The process of Claim 1, wherein the catalyst further comprises a bulk promoter.
- 23. (Original) A process for the manufacture of polycarbonate comprising reacting a feed comprising a phenol, a ketone and water in the presence of an ion exchange resin catalyst to produce an effluent comprising bisphenol; determining the *para-para* bisphenol selectivity of the reaction; adjusting the concentration of the water in the feed based upon the *para-para* bisphenol selectivity; and reacting said bisphenol with a carbonic acid derivative or a carbonate diester in the presence of a polymerization catalyst.
- 24. (Original) A process for the reaction of a ketone with a phenol to form a bisphenol in the presence of a catalyst comprising determining the catalytic activity of the catalyst over time; introducing a feed comprising water; phenol and ketone to the catalyst; reacting the phenol and ketone in the presence of the catalyst; controlling the concentration of the water in the feed based upon the amount of catalyst and catalytic activity; wherein the reaction has a para-para bisphenol selectivity of at least about 94%.

## 25. (Original) A process for the manufacture of polycarbonate comprising

synthesizing a bisphenol in the presence of a catalyst by determining the catalytic activity of the catalyst; introducing a feed comprising water, phenol and ketone to the catalyst; reacting the phenol and ketone in the presence of the catalyst to form bisphenol; controlling the concentration of the water in the feed based upon the amount of catalyst and catalytic activity; and

reacting the bisphenol with a carbonic acid derivative or a carbonic diester in the presence of a polymerization catalyst, wherein the reaction generating bisphenol has a *para-para* bisphenol selectivity of at least about 94%.

26. (Original) A process for the reaction of a ketone with a phenol to form a bisphenol comprising

reacting a feed comprising a phenol, a ketone and a first concentration of water in the presence of an ion exchange resin catalyst at a first flow rate to produce an effluent; and

reacting a feed comprising a phenol, a ketone and a second concentration of water in the presence of the ion exchange resin catalyst at a second flow rate to produce an effluent, wherein the reaction at the first flow rate has a *para-para* bisphenol selectivity within about 1% of the *para-para* bisphenol selectivity of the reaction at the second flow rate and the first concentration of water does not equal the second concentration of water.